Product: Tetrafluoromethane P-4665-F Date: December 2006

Praxair Material Safety Data Sheet

1. Chemical Product and Company Identification

Product Name: Tetrafluoromethane (MSDS No. P-4665-F)	Trade Names: Halocarbon-14
Chemical Name: Tetrafluoromethane	Synonyms: Carbon tetrafluoride, F-14, refrigerant gas R14, perfluoromethane, halon-14
Chemical Family: Halogenated alkane	Product Grades: 3.7, 4.7 semiconductor
	process gas

Telephone: Emergencies: 1-800-645-4633* Company Name: Praxair, Inc.

CHEMTREC: 1-800-424-9300* 39 Old Ridgebury Road **Routine:** 1-800-PRAXAIR Danbury, CT 06810-5113

2. Hazards Identification

EMERGENCY OVERVIEW

CAUTION! High-pressure gas.
Can cause rapid suffocation.
May cause dizziness and drowsiness.

Self-contained breathing apparatus may be required by rescue workers.

Under ambient conditions, this is a colorless odorless gas.

OSHA REGULATORY STATUS: This material is considered hazardous by the OSHA Hazard Communications Standard (29 CFR 1910.1200).

POTENTIAL HEALTH EFFECTS:

Effects of a Single (Acute) Overexposure

Inhalation. Asphyxiant. High concentrations can cause dizziness, nausea, vomiting, disorientation, confusion, incoordination, and narcosis. Very high concentrations may cause suffocation. Lack of oxygen can kill.

Skin Contact. No harm expected.

Swallowing. An unlikely route of exposure, this product is a gas at normal temperature and pressure.

Eye Contact. No harm expected.

Effects of Repeated (Chronic) Overexposure. No harm expected.

Other Effects of Overexposure. At high concentrations, tetrafluoromethane may produce cardiac arrhythmias or arrest due to sensitization of the heart to adrenaline and noradrenalin.

^{*}Call emergency numbers 24 hours a day only for spills, leaks, fire, exposure, or accidents involving this product. For routine information, contact your supplier, Praxair sales representative, or call 1-800-PRAXAIR (1-800-772-9247).

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Exposure to fluorocarbon thermal decomposition products may produce flu-like symptoms including chills, fever, weakness, muscular aches, headache, chest discomfort, sore throat, and dry cough. Complete recovery usually occurs within 24 hours after exposure.

Medical Conditions Aggravated by Overexposure. The toxicology and the physical and chemical properties of tetrafluoromethane suggest that overexposure is unlikely to aggravate existing medical conditions.

CARCINOGENICITY: Tetrafluoromethane is not listed by NTP, OSHA, or IARC.

POTENTIAL ENVIRONMENTAL EFFECTS: None known. For further information, see section 12, Ecological Information.

3. Composition/Information on Ingredients

See section 16 for important information about mixtures.

COMPONENT **CAS NUMBER** CONCENTRATION Tetrafluoromethane 75-73-0 >99%*

4. First Aid Measures

INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, qualified personnel may give oxygen. Call a physician.

SKIN CONTACT Flush with water. If discomfort persists, seek medical attention.

SWALLOWING: An unlikely route of exposure. This product is a gas at normal temperature and pressure.

EYE CONTACT: Flush eyes thoroughly with warm water. Hold the eyelids open and away from the eyeballs to ensure that all surfaces are flushed thoroughly. See a physician, preferably an ophthalmologist, immediately.

NOTES TO PHYSICIAN: Do not administer adrenaline due to the sensitizing effect of fluorocarbons on the myocardium. Treatment of overexposure should be directed at the control of symptoms and the clinical condition of the patient. Exposure to fluorocarbon pyrolosis products should be considered in the diagnostic evaluation of occupationally related fever of short duration and unknown origin. Signs of exposure include tachycardia, hyperpnea, and pharyngeal congestion: investigation may reveal pulmonary edema and leucocytosis.

5. Fire Fighting Measures

FLAMMABLE PROPERTIES: Nonflammable gas

SUITABLE EXTINGUISHING MEDIA: Tetrafluoromethane cannot catch fire. Use media appropriate for surrounding fire.

PRODUCTS OF COMBUSTION: Not applicable. At temperatures above 1832°F (1000°C), carbon dioxide will react with tetrafluoromethane to form toxic carbonyl fluoride. See section 10.

PROTECTION OF FIREFIGHTERS: CAUTION! High-Pressure gas. Asphyxiant—lack of oxygen can kill. Evacuate all personnel from danger area. Immediately deluge cylinders with water from maximum distance until cool; then move them away from fire area if without risk.

^{*}The symbol > means "greater than."

Self-contained breathing apparatus may be required by rescue workers. On-site fire brigades must comply with OSHA 29 CFR 1910.156.

Specific Physical and Chemical Hazards. Heat of fire can build pressure in cylinder and cause it to rupture. No part of cylinder should be subjected to a temperature higher than 125°F (52°C). Tetrafluoromethane cylinders are equipped with a pressure relief device. (Exceptions may exist where authorized by DOT.)

Protective Equipment and Precautions for Firefighters. Firefighters should wear self-contained breathing apparatus and full fire-fighting turnout gear.

6. Accidental Release Measures

STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED:

CAUTION! High-Pressure Gas.

Personal Precautions. Asphyxiant—lack of oxygen can kill. Evacuate all personnel from danger area. Use self-contained breathing apparatus where needed. Shut off leak if without risk. Ventilate area of leak or move cylinder to a well-ventilated area. Test for sufficient oxygen, especially in confined spaces, before allowing reentry.

Environmental Precautions. Prevent waste from contaminating the surrounding environment. Keep personnel away. Discard any product, residue, disposable container, or liner in an environmentally acceptable manner, in full compliance with federal, state, and local regulations. If necessary, call your local supplier for assistance.

7. Handling and Storage

PRECAUTIONS TO BE TAKEN IN HANDLING: Gas can cause rapid suffocation due to oxygen deficiency. Store and use with adequate ventilation. Protect cylinders from damage. Use a suitable hand truck to move cylinders; do not drag, roll, slide, or drop. Never attempt to lift a cylinder by its cap; the cap is intended solely to protect the valve. Never insert an object (e.g., wrench, screwdriver, pry bar) into cap openings; doing so may damage the valve and cause a leak. Use an adjustable strap wrench to remove over-tight or rusted caps. Open valve slowly. If valve is hard to open, discontinue use and contact your supplier. Close cylinder valve after each use; keep closed even when empty. Do not smoke in areas where fluorocarbons are used. Wash hands thoroughly after handling fluorocarbons or materials sprayed with them, especially before eating or smoking. For other precautions in using tetrafluoromethane, see section 16.

PRECAUTIONS TO BE TAKEN IN STORAGE: Store and use with adequate ventilation. **Always secure cylinders upright** to keep them from falling or being knocked over. Install cap when not in use. Screw valve protection cap firmly in place by hand. Store only where temperature will not exceed 125°F (52°C). Store full and empty cylinders separately. Use a first-in, first-out inventory system to prevent storing full cylinders for long periods.

RECOMMENDED PUBLICATIONS: For further information on storage, handling, and use, see Praxair publication P-14-153, *Guidelines for Handling Gas Cylinders and Containers*. Obtain from your local supplier.

8. Exposure Controls/Personal Protection

COMPONENT	OSHA PEL	ACGIH TLV-TWA (2006)
Tetrafluoromethane	N.E.*	N.E.*

*N.E.-Not Established.

IDLH = Not available.

ENGINEERING CONTROLS:

Local Exhaust. Use a local exhaust system, if necessary, to prevent oxygen deficiency and control the worker's exposure to high concentrations of this product.

Mechanical (General). General exhaust ventilation may be acceptable if it can maintain an adequate supply of air.

Special. None

Other. None

PERSONAL PROTECTIVE EQUIPMENT:

Skin Protection. Wear work gloves when handling cylinders. Metatarsal shoes for cylinder handling. Select in accordance with OSHA 29 CFR 1910.132 and 1910.133. Regardless of protective equipment, never touch live electrical parts.

Eye/Face Protection. Wear safety glasses when handling cylinders. Select eye protection in accordance with OSHA 29 CFR 1910.133.

Respiratory Protection. Use air-supplied respirators to protect against high concentrations of this product. In confined spaces or in oxygen-deficient atmospheres, use a full-face, self-contained breathing apparatus operated in the positive-pressure, demand mode. Respiratory protection must conform to OSHA rules as specified in 29 CFR 1910.134. Select per OSHA 29 CFR 1910.134 and ANSI Z88.2.

9. Physical and Chemical Properties **APPEARANCE:** Colorless gas ODOR: None ODOR THRESHOLD: Not applicable. PHYSICAL STATE: Gas at normal temperature and pressure Not applicable. -298.46°F (-183.59°C) **FREEZING POINT** at 1 atm: **BOILING POINT** at 101.325 kPa (1 atm): -198.51°F (-128.06°C) **FLASH POINT** (test method): Not applicable. **EVAPORATION RATE** (Butyl Acetate = 1): Not available. FLAMMABILITY: Nonflammable FLAMMABLE LIMITS IN AIR, % by volume: LOWER: Not **UPPER:** Not applicable. applicable. Not available. **VAPOR PRESSURE** at 68°F (20°C): **LIQUID DENSITY** at -112°F (-80°C): 81.28 lb/ft³ (1.302 g/cm³) 0.228 lb/ft3 (3.65 kg/m3) GAS DENSITY at 70°F (21.1°C) and 1 atm: **SPECIFIC GRAVITY** ($H_2O = 1$) at 19.4°F (-7°C): Not available.

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1	CDECIFIC CD AVITY (A:			
	SPECIFIC GRAVITY (Air = 1) at 70°F (21.1°C) and 1 atm:	3.039		
ļ	SOLUBILITY IN WATER % by wt at 77°F (25°C			
	and 1 atm:	0.0015		
	PARTITION COEFFICIENT: n-octanol/water:	1.18		
	AUTOIGNITION TEMPERATURE:	Not available.		
	DECOMPOSITION TEMPERATURE: PERCENT VOLATILES BY VOLUME:	Not available.		
	MOLECULAR WEIGHT:	88.01		
	MOLECULAR FORMULA:	CF ₄		
		·		
	10. Stability	and Reactivity		
	CHEMICAL STABILITY: ☐ Unstable ☐	Stable		
	CONDITIONS TO AVOID: None known.			
1	INCOMPATIBLE MATERIALS: Aluminum, CO ₂ above 1832°F (1000°C), alloys of more than 2% Mg in the presence of water.			
HAZARDOUS DECOMPOSITION PRODUCTS: Thermal decomposition may produce fluorine and carbonyl fluoride.				
	POSSIBILITY OF HAZARDOUS REACTIONS:		/ill Not Occur	
	Thermal decomposition may produce fluorine ar	nd carbonyl fluoride.		
	11. Toxicolo	gical Information		
	ACUTE DOSE EFFECTS: None known.			
	STUDY RESULTS: None known.			
	12. Ecologi	cal Information		
i	ECOTOXICITY: No known effects.			
	OTHER ADVERSE EFFECTS: Tetrafluorome ozone-depleting chemicals.	thane does not contain a	ny Class I or Class II	
	13. Disposa	l Considerations		
	WASTE DISPOSAL METHOD: Do not attempt	ot to dispose of residual o	or unused quantities.	
	Return cylinder to supplier.			
	14. Transp	ort Information		
	DOT/IMO SHIPPING NAME: Tetrafluoromet			
•		NTIFICATION	PRODUCT	
		MBER: UN1982	RQ: None	
	SHIPPING LABEL(s): NONFLAMMA			
l	PLACARD (when required): NONFLAMMA	BLE GAS		
1	*NA-Not applicable.			

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SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure position, in a well-ventilated vehicle. Cylinders transported in an enclosed, nonventilated compartment of a vehicle can present serious safety hazards.

Shipment of compressed gas cylinders that have been filled without the owner's consent is a violation of federal law [49 CFR 173.301(b)].

MARINE POLLUTANTS: Tetrafluoromethane is not listed as a marine pollutant by DOT.

15. Regulatory Information

The following selected regulatory requirements may apply to this product. Not all such requirements are identified. Users of this product are solely responsible for compliance with all applicable federal, state, and local regulations.

U.S. FEDERAL REGULATIONS:

EPA (ENVIRONMENTAL PROTECTION AGENCY)

CERCLA: COMPREHENSIVE ENVIRONMENTAL RESPONSE, COMPENSATION, AND LIABILITY ACT OF 1980 (40 CFR Parts 117 and 302):

Reportable Quantity (RQ): None

SARA: SUPERFUND AMENDMENT AND REAUTHORIZATION ACT:

SECTIONS 302/304: Require emergency planning based on Threshold Planning Quantity (TPQ) and release reporting based on Reportable Quantities (RQ) of Extremely Hazardous Substances (EHS) (40 CFR Part 355):

TPQ: None

EHS RQ (40 CFR 355): None

SECTIONS 311/312: Require submission of MSDSs and reporting of chemical inventories with identification of EPA hazard categories. The hazard categories for this product are as follows:

IMMEDIATE: Yes PRESSURE: Yes DELAYED: No REACTIVITY: No

FIRE: No

SECTION 313: Requires submission of annual reports of release of toxic chemicals that appear in 40 CFR Part 372.

Tetrafluoromethane is not subject to reporting under Section 313.

40 CFR 68: RISK MANAGEMENT PROGRAM FOR CHEMICAL ACCIDENTAL RELEASE PREVENTION: Requires development and implementation of risk management programs at facilities that manufacture, use, store, or otherwise handle regulated substances in quantities that exceed specified thresholds.

Tetrafluoromethane is not listed as a regulated substance.

TSCA: TOXIC SUBSTANCES CONTROL ACT: Tetrafluoromethane is listed on the TSCA inventory.

OSHA: OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION:

29 CFR 1910.119: PROCESS SAFETY MANAGEMENT OF HIGHLY HAZARDOUS CHEMICALS: Requires facilities to develop a process safety management program based on Threshold Quantities (TQ) of highly hazardous chemicals.

Tetrafluoromethane is not listed in Appendix A as a highly hazardous chemical.

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STATE REGULATIONS:

CALIFORNIA: Tetrafluoromethane is not listed by California under the SAFE DRINKING WATER AND TOXIC ENFORCEMENT ACT OF 1986 (Proposition 65).

PENNSYLVANIA: Tetrafluoromethane is subject to the PENNSYLVANIA WORKER AND COMMUNITY RIGHT-TO-KNOW ACT (35 P.S. Sections 7301-7320).

16. Other Information

Be sure to read and understand all labels and instructions supplied with all containers of this product.

OTHER HAZARDOUS CONDITIONS OF HANDLING, STORAGE, AND USE: High-pressure gas. Use piping and equipment adequately designed to withstand pressures to be encountered. Use only in a closed system. Use a backflow prevention device in the piping. Never work on a pressurized system. If there is a leak, close the cylinder valve. Blow the system down in a safe and environmentally sound manner in compliance with all federal, state, and local laws; then repair the leak. Follow safe practices when returning cylinder to supplier. Be sure valve is closed; then install valve outlet cap or plug, leak-tight. Never place a compressed gas cylinder where it may become part of an electrical circuit.

NOTE: Prior to using any plastics, confirm their compatibility with tetrafluoromethane.

Mixtures. When you mix two or more gases, you can create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an industrial hygienist or other trained person when you evaluate the end product. Remember, gases have properties that can cause serious injury or death.

HAZARD RATING SYSTEMS:

NFPA RATINGS:		HMIS RATINGS:	
HEALTH	= 1	HEALTH	= 0
FLAMMABILITY	= 0	FLAMMABILITY	= 0
INSTABILITY	= 0	PHYSICAL HAZARD	= 3
SPECIAL	= None		

STANDARD VALVE CONNECTIONS FOR U.S. AND CANADA:

THREADED: CGA-580, CGA-320 PIN-INDEXED YOKE: Not applicable. ULTRA-HIGH-INTEGRITY CONNECTION: CGA-716

Use the proper CGA connections. **DO NOT USE ADAPTERS.** Additional limited-standard connections may apply. See CGA pamphlet V-1 listed below.

Ask your supplier about free Praxair safety literature as referred to in this MSDS and on the label for this product. Further information can be found in the following materials published by the Compressed Gas Association, Inc. (CGA), 4221 Walney Road, 5th Floor, Chantilly, VA 20151-2923, Telephone (703) 788-2700, http://www.cganet.com/Publication.asp.

AV-1	Safe Handling and Storage of Compressed Gases
P-1	Safe Handling of Compressed Gases in Containers
SB-2	Oxygen-Deficient Atmospheres
V-1	Compressed Gas Cylinder Valve Inlet and Outlet Connections
_	Handbook of Compressed Gases. Fourth Edition

Praxair asks users of this product to study this MSDS and become aware of product hazards and safety information. To promote safe use of this product, a user should (1) notify employees, agents, and contractors of the information in this MSDS and of any other known product hazards and safety information, (2) furnish this information to each purchaser of the product, and (3) ask each purchaser to notify its employees and customers of the product hazards and safety information.

The opinions expressed herein are those of qualified experts within Praxair, Inc. We believe that the information contained herein is current as of the date of this Material Safety Data Sheet. Since the use of this information and the conditions of use of the product are not within the control of Praxair, Inc., it is the user's obligation to determine the conditions of safe use of the product.

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